## **Summary and Comparison of Liquid Disinfectants**

Class	Recommended Use	How They Work	Advantages	Disadvantages	Comments & Hazards	Examples
70% Isopropyl Alcohol Solution	<ul> <li>Cleaning some instruments</li> <li>Cleaning skin</li> </ul>	<ul> <li>Changes protein structure of microorganism</li> <li>Presence of water assists with killing action</li> </ul>	■ Fairly inexpensive	<ul> <li>&lt;50% solution not very effective</li> <li>Not active when organic matter present</li> <li>Not active against certain types of viruses</li> <li>Evaporates quickly</li> <li>Contact time not sufficient for killing</li> </ul>	<ul><li>Flammable</li><li>Eye irritant</li><li>Toxic</li></ul>	
Chlorine Compounds	<ul> <li>Spills of human body fluids</li> <li>Bactericidal – good</li> <li>Fungicidal – good</li> <li>Sporicidal – good at &gt; 1000 ppm Sodium Hypochlorite</li> </ul>	<ul> <li>Free available chlorine combines with contents within microorganism, reaction byproducts cause its death</li> <li>Need 500 to 5000 ppm</li> <li>Produce chemical combination with cell substances</li> <li>Depends upon release of hypochlorous acid</li> </ul>	<ul> <li>Kills hardy viruses         (e.g., hepatitis)</li> <li>Kills a wide range of         organisms</li> <li>Inexpensive</li> <li>Penetrates well</li> <li>Relatively quick         microbial kill</li> <li>May be used on food         prep surfaces</li> </ul>	<ul> <li>Corrodes metals such as stainless, aluminum</li> <li>Organics may reduce activity</li> <li>Increase in alkalinity decreases bactericidal property</li> <li>Unpleasant taste and odor</li> <li>Tuberculocidal, with extended contact time</li> </ul>	<ul> <li>Follow spill procedure and dilution instructions</li> <li>Make fresh solutions before use</li> <li>Eye, skin and respiratory irritant</li> <li>Corrosive</li> <li>Toxic</li> </ul>	<ul> <li>Bleach solutions (sodium hypochlorite)</li> <li>Clorox</li> <li>Cyosan</li> <li>Purex</li> </ul>
Glutaraldehyde	<ul> <li>Bactericidal – good</li> <li>Fungicidal – good</li> <li>Tuberculocidal – excellent</li> <li>Virucidal – good</li> <li>Sporicidal – good</li> </ul>	Coagulates cellular proteins	<ul> <li>Non-staining, relatively noncorrosive</li> <li>Useable as a sterilant on plastics, rubber, lenses, stainless steel and other items that can't be autoclaved</li> </ul>	<ul> <li>Not stable in solution</li> <li>Has to be in alkaline solution</li> <li>Inactivated by organic material</li> </ul>	<ul> <li>Eye, skin and respiratory irritant</li> <li>Sensitizer</li> <li>Toxic</li> </ul>	<ul><li>Calgocide 14</li><li>Cidex</li><li>Vespore</li></ul>

Class	Recommended Use	How They Work	Advantages	Disadvantages	Comments & Hazards	Examples
Iodophors (Iodine with carrier)	<ul> <li>Disinfecting some semicritical medical equipment</li> <li>Bactericidal – very good</li> <li>Fungicidal – excellent</li> <li>Virucidal - excellent</li> </ul>	<ul> <li>Free iodine enters microorganism and binds with cellular components</li> <li>Carrier helps penetrate soil/fat</li> <li>Need 30 to 50 ppm</li> <li>Probably by disorder of protein synthesis due to hindrance and/or blocking of hydrogen bonding</li> </ul>	<ul> <li>Kills broad range of organisms</li> <li>Highly reactive</li> <li>Low tissue toxicity</li> <li>Kills immediately rather than by prolonged period of stasis</li> <li>Not affected by hard water</li> <li>May be used on food prep surfaces</li> </ul>	<ul> <li>May stain plastics or corrode metal</li> <li>May stain skin/ laundry</li> <li>Stains most materials</li> <li>Odor</li> <li>Some organic and inorganic substances neutralize effect</li> <li>Tuberculocidal, with extended contact time</li> <li>Sporicidal, some</li> </ul>	<ul> <li>Dilution critical</li> <li>Follow directions!</li> <li>Use only EPA         registered hard         surface iodophor         disinfectants</li> <li>Don't confuse skin         antiseptic iodophors         for disinfectants</li> <li>Skin and eye irritant</li> <li>Corrosive</li> <li>Toxic</li> </ul>	Bactergent Hy-Sine loprep Providone (iodine /betadine) Wescodyne
Phenolic Compounds	<ul> <li>Bactericidal – excellent</li> <li>Fungicidal – excellent</li> <li>Tuberculocidal – excellent</li> <li>Virucidal – excellent</li> </ul>	<ul> <li>Gross protoplasmic poison</li> <li>Disrupts cell walls</li> <li>Precipitates cell proteins</li> <li>Low concentrations inactivate essential enzyme systems</li> </ul>	<ul> <li>Nonspecific concerning bactericidal and fungicidal action</li> <li>When boiling water would cause rusting, the presence of phenolic substances produces an antirusting effect</li> </ul>	<ul> <li>Unpleasant odor</li> <li>Some areas have disposal restrictions</li> <li>Effectiveness reduced by alkaline pH, natural soap or organic material</li> <li>Sporicidal, no</li> </ul>	<ul> <li>Skin and eye irritant</li> <li>Sensitizer</li> <li>Corrosive</li> <li>Toxic</li> </ul>	<ul><li>Hil-Phene</li><li>Lph</li><li>Metar</li><li>Vesphene</li></ul>
Quaternary Ammonium Compounds (QUATS)	<ul> <li>Ordinary housekeeping (e.g., floors, furniture, walls)</li> <li>Bactericidal – excellent</li> <li>Fungicidal – good</li> <li>Virucidal – good (not as effective as phenols)</li> </ul>	<ul> <li>Affects proteins and cell membrane of microorganism</li> <li>Releases nitrogen and phosphorous from cells</li> </ul>	<ul> <li>Contains a detergent to help loosen soil</li> <li>Rapid action</li> <li>Colorless, odorless</li> <li>Non-toxic, less corrosive</li> <li>Highly stable</li> <li>May be used on food prep surfaces</li> </ul>	<ul> <li>Does not eliminate spores, TB bacteria, some viruses</li> <li>Effectiveness influenced by hard water</li> <li>Layer of soap interferes with action</li> </ul>	<ul> <li>Select from EPA list of hospital disinfectants</li> <li>Skin and eye irritant</li> <li>Toxic</li> </ul>	<ul><li>Coverage 258</li><li>End-Bac</li><li>Hi Tor</li></ul>

This information was provided to the University of Virginia by:

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