

# Powder Data Analysis: Indexing & Structural Refinement based on the Rietveld Method

An introduction to the basic knowledge required for Project 2

X-ray Crystallography

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THE UNIVERSITY OF  
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# Analyzing Powder Data of New Material

- Assumption: A high crystallinity, phase-pure sample has been prepared, and good data have been collected
  - Refer to Powder Methods handout for sample prep and data collection!
- First step is generally a search & match of a powder pattern database
  - ICDD's PDF (Powder Diffraction File) if you have access
  - COD (Crystallography Open Database) allows export of a PDF as well
    - <http://www.crystallography.net/cod/>
- If you find a reasonable match: Does the composition found make sense for the compound you measured?
  - E.g., AlScMo<sub>3</sub>O<sub>12</sub> gave a match to Fe<sub>2</sub>Mo<sub>3</sub>O<sub>12</sub>, which makes sense
  - “Substitute” atoms and use as starting structural model
- If no match is found: You need to find a cell by *indexing*

# Indexing of Powder Data

- **Indexing of powder data is much harder than one may think**
  - Need to determine 6 or less variables (lattice parameters)
- **Problem: Only 1D data are available**
  - We are trying to solve a 3D problem
  - Peak overlap (exact or partial)
  - Weak peaks and systematic absences
  - Random and systematic errors
  - Usually more than 1 unique solution possible
- **We are trying to solve the following equation:**

$$\frac{1}{d^2} = h^2a^{*2} + k^2b^{*2} + l^2c^{*2} + 2hk a^*b^*\cos\gamma^* + 2hl a^*c^*\cos\beta^* + 2kl b^*c^*\cos\alpha^*$$

$$\frac{1}{d^2} = h^2\mathbf{A} + k^2\mathbf{B} + l^2\mathbf{C} + hk \mathbf{D} + hl \mathbf{E} + kl \mathbf{F}$$



# Indexing Approaches

- There are several ways to solve the equation on the previous slide
- The most common algorithms are referred to as
  - Zone indexing (used in ITO)
  - Successive dichotomy (used in DICVOL – exhaustive search)
  - Index heuristics (used in TREOR – “smart trial and error”)
  - There are also some more modern routines that are implemented in modern Rietveld codes
- We will not talk about the underlying equations/details for each of these algorithms, but Project 2 will involve indexing powder data
- It can be a good idea to use multiple indexing programs for the same data
  - When one fails, another one may succeed
  - If we get the same cell with multiple programs, it increases our confidence in that cell

# **Indexing – Practical Considerations**

- **In contrast to automated indexing algorithms implemented in single crystal programs, YOU decide what peaks to include in indexing powder data!**
  - As a famous powder crystallographer has said, "Even the best program will not help you if you give it bad data!"
- **Two main options:**
  - Manual peak fitting, choose peaks from output to include in indexing list
  - Use indexing program that starts with powder pattern
- **It may be wise to exclude unreliable peaks (weak, overlapping)**
- **Closely inspect output**
  - Figure of merit is used to rank cells – the higher, the better
  - However, more important question is whether all strong/reliable peaks are accounted for!

# Successfully Indexed – What's Next?

- **Once you have identified a cell that agrees well with your indexing peak list, you can confirm it by doing a Le Bail or Pawley fit**
  - Requires unit cell constants and space group, but no structural model
  - Look at systematic absences to choose a reasonable space group
  - Intensities are considered variables
  - Inspect your difference curve: Are all peaks accounted for?
- **Extracted intensities can be exported and used for structure solution**
  - We will not get into this in this class, as it is non-trivial!
- **Le Bail or Pawley fit shows what the best possible fit using a structural model could look like**
- **Once you have a structural starting model (from any source – search/match, structure solution etc.):**

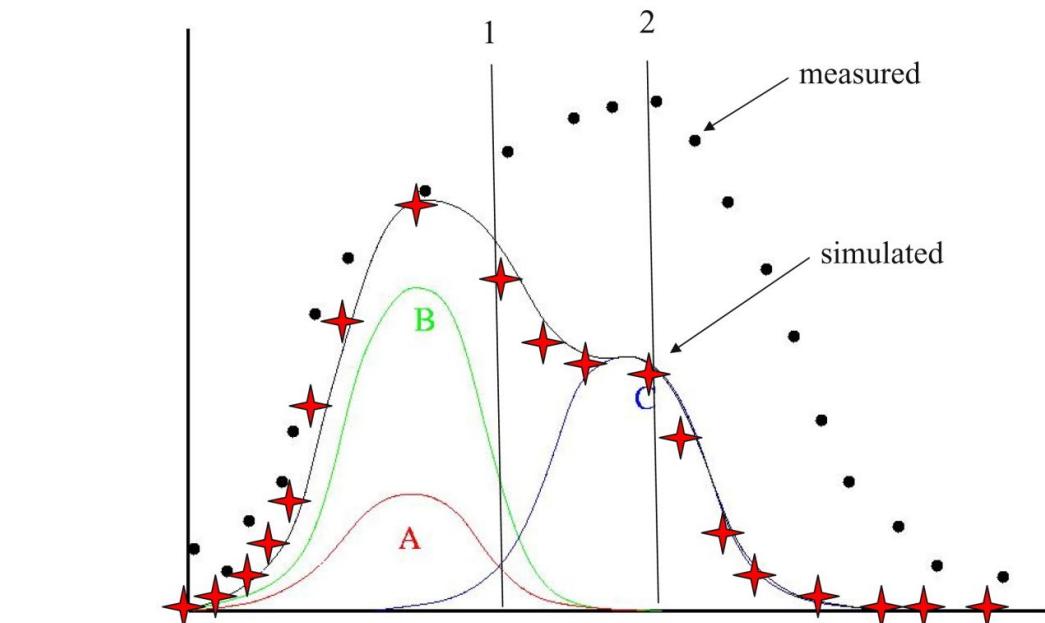
Carry out a Rietveld refinement!

# Powder crystallography before Rietveld

- Primary strategy: AVOID when it comes to structure determination!
  - There was no straightforward way to deal with data
  - Had to manually integrate intensities
  - Overlapping reflections were a big problem
    - Usually discarded
    - Alternative: Rewriting of single crystal software to refine using sums of overlapped reflections
- Powder pattern simulation was more common
  - Relatively straightforward
  - Conclusions drawn based on similarities between patterns (e.g., isostructural compounds)
    - Visual comparison
- Main use of powder diffraction was for phase identification

# Major breakthrough: Overlapped reflections

- Rietveld witnessed the power of introducing computers into crystallography during his dissertation (1961-1964)
- He realized that computers can handle individual datapoint intensities, which allowed calculation of  $|F_{hkl}|$  even for overlapped reflection!



# History of the Rietveld method

- **First introduced by Hugo Rietveld in 1966-1969**
  - He realized that if a pattern can be simulated/calculated, it should be possible to optimize the agreement between the calculated pattern and the collected data
  - Little response from the scientific community after first presentation of the method in 1966
  - Shared his program as well as knowledge freely
- **Became possible by increase in computing power**
  - Nowadays, a number of program codes are freely available for a variety of operating systems
- **Originally introduced for neutron data (Rietveld worked at a neutron reactor), first used for X-ray data in 1977**

# What is the Rietveld method?

- **Least squares based minimization algorithm to obtain the best fit between a structural model and a powder pattern**
  - Demanding, as the algorithm is non-linear
  - User decides which parts of the model can be varied
- **Each point in the pattern can be regarded as an observation**
  - “No Bragg intensity” tells you something about your material, too!
- **Full pattern fitting**
  - In contrast to single crystal data, “experiment dependent parameters” must be fitted as well: Background, peak shape – sample and instrument contributions, lattice constants, ...
- **Requires an approximate starting model**
  - Rietveld can NOT be used to “solve structures”, it is a refinement method!

# Goal of Rietveld refinement

- Obtain the best possible fit between a structural model and a powder pattern
  - User decides what parameters can be varied
- Can be used to confirm that a sample has a specific structure
  - Especially for single phase samples, a good fit is usually accepted as confirmation of phase purity/structural identity
- Useful for quantitative analysis
  - This is by definition a multi-phase refinement!
- Frequently used to investigate structural changes as a response to external stimuli
  - Thermal expansion (lattice constants as  $f(T)$ ), compressibility (lattice constants as  $f(P)$ ), oxygen deficiency (site occupancy as  $f(P(O_2))$  etc.

# Calculated powder patterns

- **Steps involved in calculating a powder pattern:**
  - Generate list of reflections
    - From unit cell data and space group
  - Compute  $F_{hkl}$ 
    - Requires atomic positions from model
  - Generate peak heights from  $|F_{hkl}|^2 \cdot \text{multiplicity}$
  - Convolute peaks
    - Peak shape function
  - Add background
    - Could be a constant (rare), simple function (most common), or a function that accounts for amorphous materials' contributions
- **Rietveld: Optimize parameters to match experimental data**

# Possibilities

- **Works for simple and complicated structures**
  - Thanks to today's computing power, even complicated structures can be refined rather rapidly
- **Can be used to refine several phases as well as mixed occupancies**
  - Use of internal standard possible
  - Quantitative analysis of mixture or versus a standard
  - Non-stoichiometry/partial occupancy can be refined
- **Each point in the pattern can be regarded as an observation**
  - "No Bragg intensity" tells you something about your crystal, too!
- **Refinement of several data sets together**
  - X-ray and neutron data
  - Several different wavelengths => changes scattering contrast between atoms



# Limitations

- **Parameters can sometimes be correlated**
  - e.g. zero point and sample height
- **For big structures, constraints or restraints can be necessary**
  - Restrain bond distances or bond angles (define rigid bodies)
  - Constrain composition if known
- **The method only works if you have a good starting model!**
  - Divergence might be observed
  - A local instead of a global minimum may be found

# The most important part: Data collection!

- **Without high quality data, even the best program can't help you!**
- **Choice of instrument**
  - Resolution
  - Accessible angular range
  - "Tricks" you can play
- **Sample preparation (powder methods handout!)**
  - Particle size
  - Surface roughness
  - Homogeneity
  - Preferred orientation
- **Instrument alignment, sample height and other errors**

# Choosing a Rietveld program

- Let's assume that you have access to a good diffractometer, prepared an excellent sample, chose an appropriate setup, and thus collected refinement quality data. So, what's next?
- You have to choose a Rietveld program, export your data in the appropriate format, and set up an input file for your refinement
  - Most diffractometer software gives you multiple export options
  - We will use GSAS-II in this class
- But how do you decide which Rietveld program you should use? And does it matter?

# First Consideration: Commercial or Freeware?

## ■ Pros

- Freeware:
  - Well, it's free!
  - You can use it anywhere
  - It usually deals well with data from many different sources
- Commercial:
  - If it came with the instrument you are using, it is probably very well programmed to handle data from that instrument!
  - You usually have access to (paid, professional) customer support & documentation
  - Especially in recent years, some of the codes have become very advanced

## ■ Cons

- Freeware:
  - No professional customer support, you have to often figure things out on your own
  - Code may have limitations, and people may or may not care about your concerns!
- Commercial:
  - Comes with a price tag
  - If you move, you may no longer have access to a program & need to learn another one



# A partial list of Rietveld programs

- **There are quite a few Rietveld programs out there!**
  - GSAS, GSAS-II
  - FullProf
  - Rietan
  - BGMN
  - Highscore Plus
  - PDXL
  - Topas, Topas Academic
  - Jade Plus
- **This is not a comprehensive list – and Rietveld codes are also included in many structure solution codes like EXPO, FOX etc.**

# Least squares minimization

- All Rietveld programs try to minimize the following sum:

$$\sum w_i [y_i - Y(x_i, \mathbf{p})]^2 \text{ where}$$

$y_i$  = observed powder diffraction intensities

Data

$Y(x_i, \mathbf{p})$  = computed diffraction intensities

Model

from  $F_{hkl}(\text{calc})$ , background model, profile convolution, correction terms...

Parameters

$\mathbf{p}$  = parameters

# Parameters

- **Structural variables**
  - Atom positions, fractional occupancies, atomic displacement parameters (ADPs)
  - Only these parameters are refinable in most single crystal software
- **Profile parameters**
  - Background
  - Peak shape, including width and asymmetry
  - Unit cell constants
  - Wavelength
  - Diffractometer zero point
  - Sample height and transparency
- **Correction terms**
  - Absorption
  - Extinction
  - Surface roughness
  - Preferred orientation

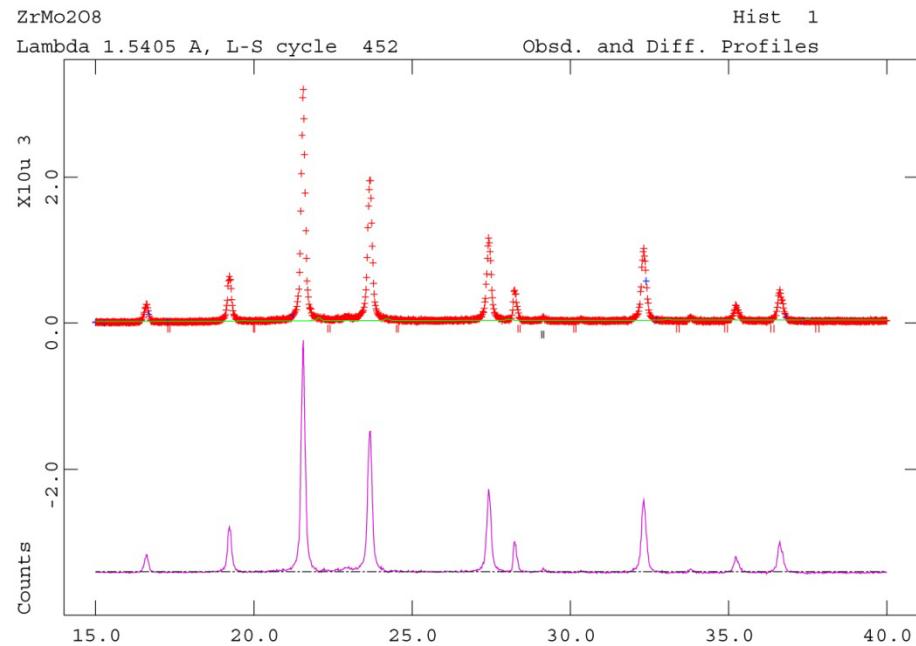
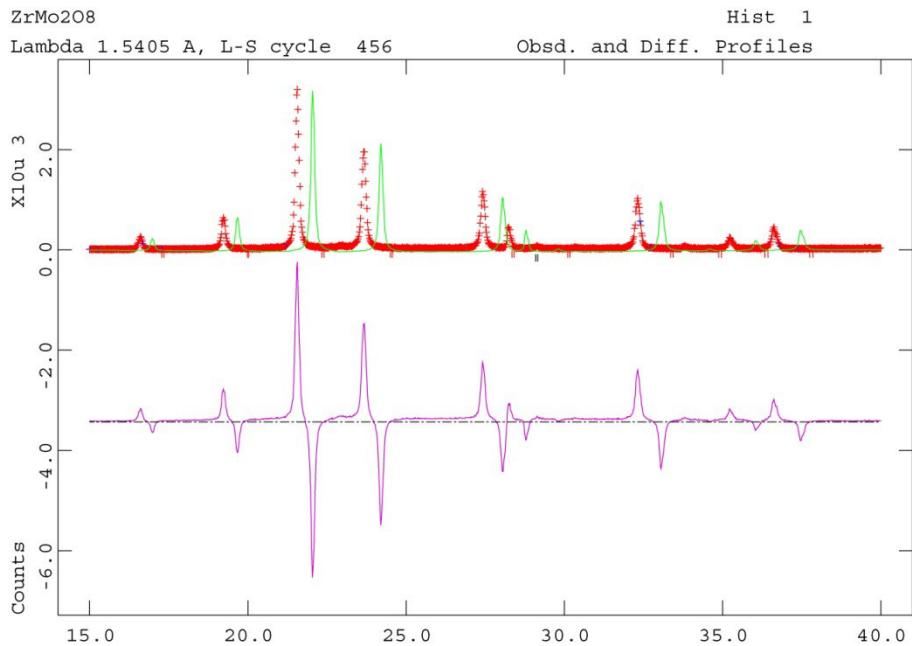
# Order of refining parameters

- In the vast majority of cases, variables have to be added to a Rietveld refinement in a stepwise fashion to avoid divergence
  - Program may otherwise try to account for some features by changing the wrong variable!
- There are rules of thumb for the order of turning on variables, however, there is NO “one-fits-all” approach!
  - Inspect your difference curve to identify what causes the worst discrepancy – this variable should be tackled first!
- Generally, the structure (atom positions, occupancies, ADPs) are varied last
- Make sure background, peak positions, and approximate peak shapes are accounted for before attempting to vary atom positions
  - Note that none of these are “variables” in single crystal refinements!

# First things first – peak positions

- **Several variables affect peak positions**
  - Wavelength (unlikely culprit for lab instruments)
  - Zero error (usually small)
  - Sample displacement/transparency (usually small)
  - Lattice constants
- **Make sure starting lattice constants are close to what they should be**
  - Often obtained from PDF search/match
  - May require indexing – either from scratch or “manually” (from known structure/indices)
- **Calculated peak positions must be close to observed peaks for a refinement to work!**
  - Otherwise, a “flat line fit” may result

# First things first - peak positions

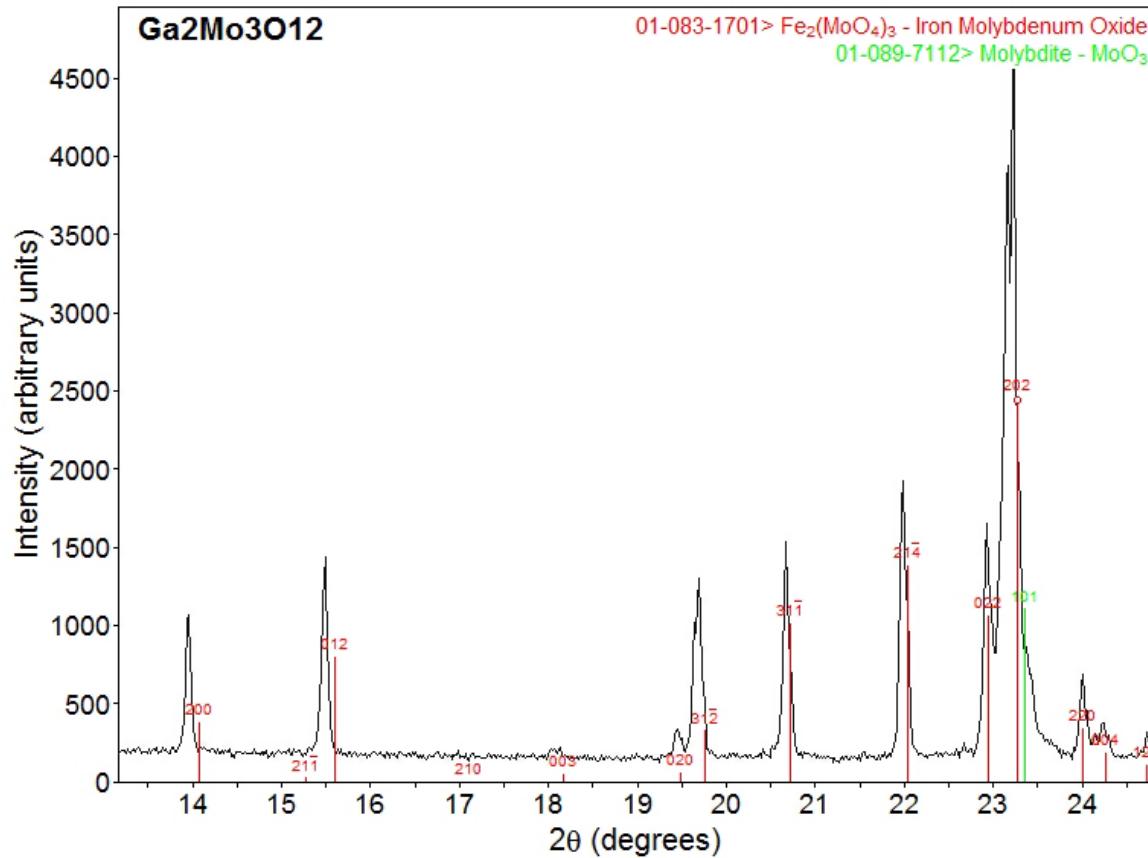


2 0 2 3 - 2 0 2 4

2 3



# “Manual indexing”



2 0 2 3 - 2 0 2 4

2 4

# “Manual indexing”

TABLE 2.4. Expressions for  $d^*(hkl)$  and  $d(hkl)$  in the Seven Crystal Systems<sup>a</sup>

System	$d^{*2}(hkl)$	$d^2(hkl)$
Triclinic	$h^2a^{*2} + k^2b^{*2} + l^2c^{*2} + 2klb^*c^*\cos\alpha^* + 2lhc^*a^*\cos\beta^* + 2hka^*b^*\cos\gamma^*$	$K^2/d^{*2}(hkl)$
Monoclinic	$h^2a^{*2} + k^2b^{*2} + l^2c^{*2} + 2hla^*c^*\cos\beta^*$	$\left\{ \frac{1}{\sin^2\beta} \left[ \frac{h^2}{a^2} + \frac{l^2}{c^2} - \frac{2hl\cos\beta}{ac} \right] + \frac{k^2}{b^2} \right\}^{-1}$
Orthorhombic	$h^2a^{*2} + k^2b^{*2} + l^2c^{*2}$	$\left\{ \frac{h^2}{a^2} + \frac{k^2}{b^2} + \frac{l^2}{c^2} \right\}^{-1}$
Tetragonal	$(h^2 + k^2)a^{*2} + l^2c^{*2}$	$\left\{ \frac{h^2 + k^2}{a^2} + \frac{l^2}{c^2} \right\}^{-1}$
Hexagonal and trigonal ( <i>P</i> )	$(h^2 + k^2 + hk)a^{*2} + l^2c^{*2}$	$\left\{ \frac{4(h^2 + k^2 + hk)}{3a^2} + \frac{l^2}{c^2} \right\}^{-1}$
Trigonal ( <i>R</i> ) (rhombohedral)	$[h^2 + k^2 + l^2 + 2(hk + kl + hl)(\cos\alpha^*)]a^{*2}$	$a^2(TR)^{-1}, \text{ where } T = h^2 + k^2 + l^2 + 2(hk + kl + hl)[(\cos^2\alpha - \cos\alpha)/\sin^2\alpha] \text{ and } R = (\sin^2\alpha)/(1 - 3\cos^2\alpha + 2\cos^3\alpha)$
Cubic	$(h^2 + k^2 + l^2)a^{*2}$	$\left\{ \frac{h^2 + k^2 + l^2}{a^2} \right\}^{-1} = \frac{a^2}{h^2 + k^2 + l^2}$

<sup>a</sup> In the monoclinic system,  $d(100) = a \sin\beta$ ,  $d(001) = c \sin\beta$ , and hence  $a = K/(a^* \sin\beta^*)$  and  $c = K/(c^* \sin\beta^*)$ .

In the hexagonal system (and trigonal *P*),  $a = b = K/(a^* \sin\gamma^*) = K/(a^* \sqrt{3}/2)$ .

In general, the expressions for  $d^{*2}$  are simpler in form than the corresponding expressions for  $d^2$ .

“Structure Determination by X-ray Crystallography”, Ladd and Palmer, Plenum, 1994.

# Errors/parameters that affect peak positions

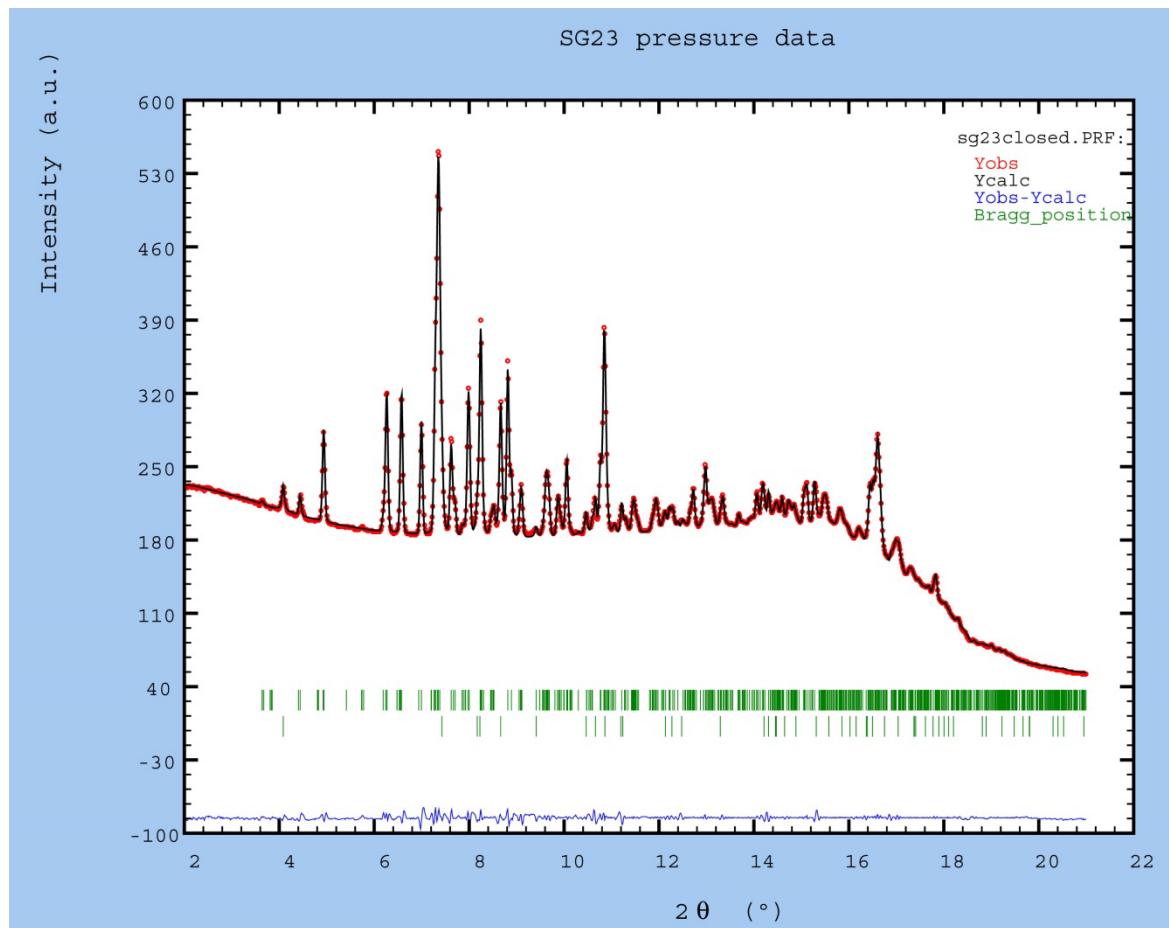
- **Wavelength (fixed for lab instrument!)**
- **Zero error – check with standard**
- **Sample displacement and transparency**
  - Transparency is effectively a “sample displacement” of part of the sample!
  - Common error for Bragg-Brentano setup
  - Do NOT vary for parallel beam setup or capillary experiments!
- **Note that sample displacement and zero error are usually strongly correlated!**
  - Unless you have a large data range, excellent data, and an internal standard, do NOT attempt to refine both simultaneously!
- **Again: YOU decide what should reasonably be varied**

# Important early variable: Background

- Unless you spot a significant other “biggest issue”, make sure your background is reasonable before refining peak shapes or structure!
- Background can be determined manually
  - May be necessary for significant background contributions from complex experimental setups or amorphous sample components
  - Program uses x,y pairs given by user and linear interpolation
- Background can be calculated using a number of mathematical functions
  - Available in most codes: Fourier cosine series, radial distribution functions, Chebychev polynomial
- For a calculated background, the number of background terms can be chosen by user
  - Usually start with 3 to 6, adjust as necessary – adjustments could be done later as background errors become more obvious

# Patterns requiring manual background

- High pressure data collected inside a diamond anvil cell – scattering from diamonds, pressure transmitting liquid etc.



# Approaches to peak shape models

- **Empirical models**
  - Functions are chosen because they are simple and fit the peaks
  - Parameters cannot be interpreted as there is no relationship to underlying physics
- **Physically-based models**
  - Functions are based on physical phenomena
  - Often found empirically originally, but parameters have meaning
- **Fundamental parameter approach**
  - Functions and even many parameter values are determined based on the underlying physics
  - Only sample property related parameters are refined

# Peak shape models

- Several different peak shape models are available in most Rietveld packages, as the peak shape is determined by
  - sample characteristics (size, strain)
  - instrument characteristics
- Convolution of several different factors, but in many cases, one factor will be dominant
- Laboratory x-rays will usually give Lorentzian peak shape
- Medium resolution CW neutron diffractometers give Gaussian peak shape
- Neutron TOF instruments give highly asymmetric peak shapes
- In high resolution data, sample characteristics determine the peak shape
  - can be very challenging to model the peak shape properly



# Physics behind peak shapes

## Common factors

- Instrument resolution (intrinsic peak width due to instrument)
- Axial divergence (low angle asymmetry)
- Sample position, transparency
- Crystallite size broadening\*
- Strain broadening\*

## Less common factors<sup>#</sup>

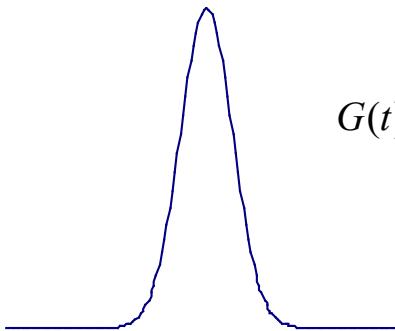
- Stacking faults
- Modulated structures
- Compositional inhomogeneity

\* Could be anisotropic (only affects certain classes of reflections then)

# Often hard or impossible to model

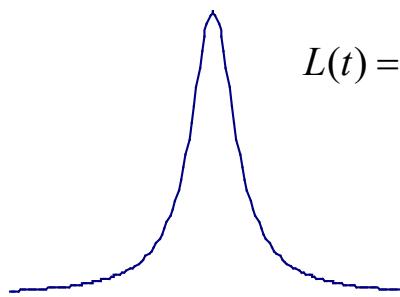
# Examples of peak shapes

Gaussian peak shape



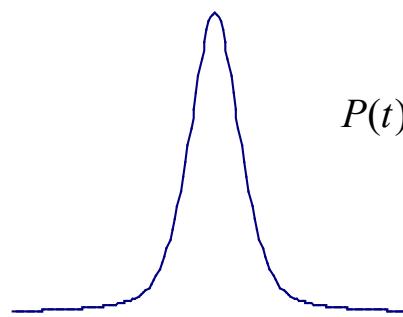
$$G(t) = \frac{1}{\sqrt{2\pi\sigma^2}} \exp\left(\frac{-t^2}{2\sigma^2}\right)$$

Lorentzian peak shape



$$L(t) = \frac{\Gamma}{2\pi} \frac{1}{(\Gamma/2)^2 + t^2}$$

Pseudo-Voigt peak shape



$$P(t) = \eta \cdot L(t, \Gamma) + (1 - \eta) \cdot G(t, \sigma)$$

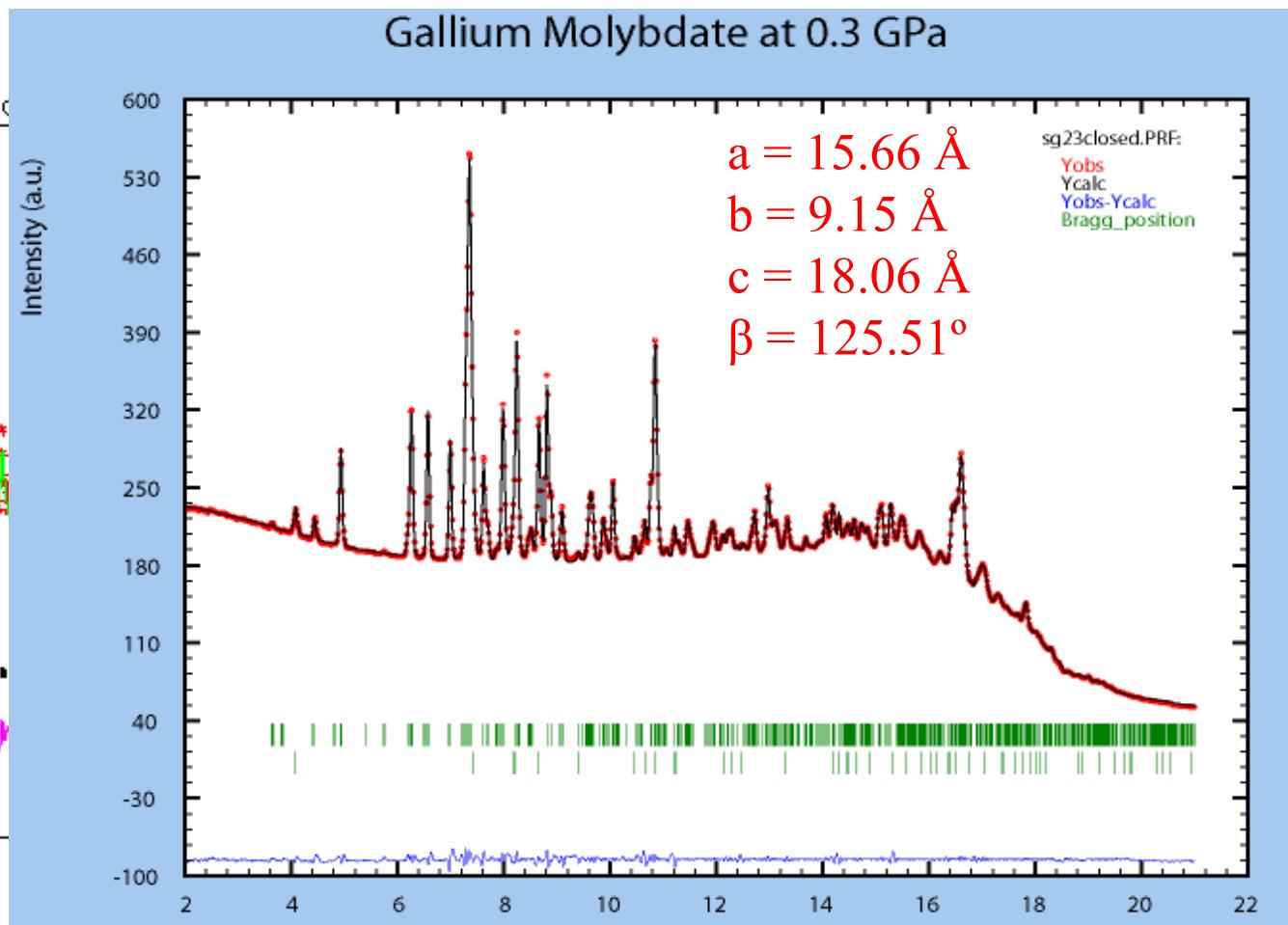
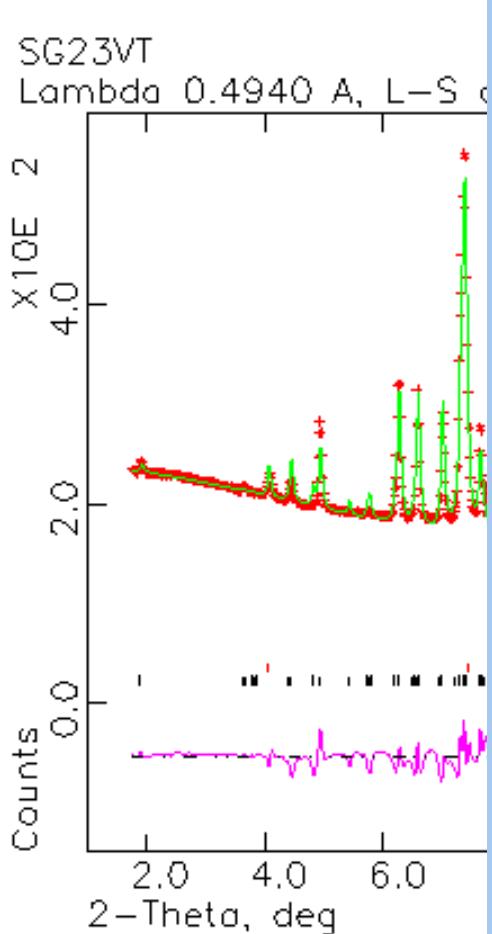
# Finally – getting ready to vary the structure!

- Before you start varying your structural model, closely inspect your refinement/difference curve!
  - Is background properly fitted? Is the peak shape reasonable?
  - Are all peaks accounted for (tick marks) and in the right positions?
- Consider your data quality and the complexity of the structure
  - Can you expect a stable refinement if you independently vary all parameters?
    - For lab X-ray data, the answer is often “no”!
  - Would constraints or restraints make sense?
    - Now is the time to implement them before your structure may fall apart during refinement!
    - Composition, thermal parameters, atomic distances, rigid bodies...
  - Look carefully at your model
    - e.g., refining the positional parameters of a half-occupied site just off an inversion center (due to disorder) will likely cause trouble
    - In some programs, special positions will still vary if you release their positional parameters!

# What if...

- ...you did everything you could to fulfill the demands on the previous slides but things still don't look as good as you would like them to
- If you start varying “non-standard variables” (correction terms), make a backup of your current input file – could save you A LOT of time!
  - Depending on program, copy the whole folder. GSAS makes backup copies automatically, so no need to worry about this!
- If the major problem lies in judging intensity mismatches, consider doing a Le Bail or Pawley refinement
  - Structureless refinement, intensities are considered variables
  - Can be useful to get a good idea of “best possible fit”, too
  - Could show problems due to incorrect space group assignment
    - missing peaks or lots of peak tick marks with zero calculated intensity
  - Also used in SDPD to extract integrated intensities from powder data

# Structural fit versus Le Bail refinement



# Varying the structure

- For lab data, never attempt to refine anisotropic displacement parameters!
  - Usually reserved for high quality neutron data
  - Occasionally attempted for synchrotron data
- For materials with mixed site occupancy, constrain positional parameters and ADPs for the two element sites!
- For lab X-ray data, it is quite common to only refine one ADP value per atom type (or possibly one for all atoms in a phase!)
  - Again, use constraints
- Damping of variables can help or hinder refinement
- Usually start with ADPs, then positions of heavy atoms, then positions of light atoms, occupancies last (if necessary, at all – do not vary if known!)
  - Watch evolution of refinement/statistics!
  - Go back to previous copy if necessary!
  - Restraints on bond distances can help stabilize structures during refinement!

# How to judge your refinement

- Several different indices:

- $R_{wp} = \left[ \frac{\sum w_i (y_i(obs) - y_i(calc))^2}{\sum w_i (y_i(obs))^2} \right]^{1/2}$

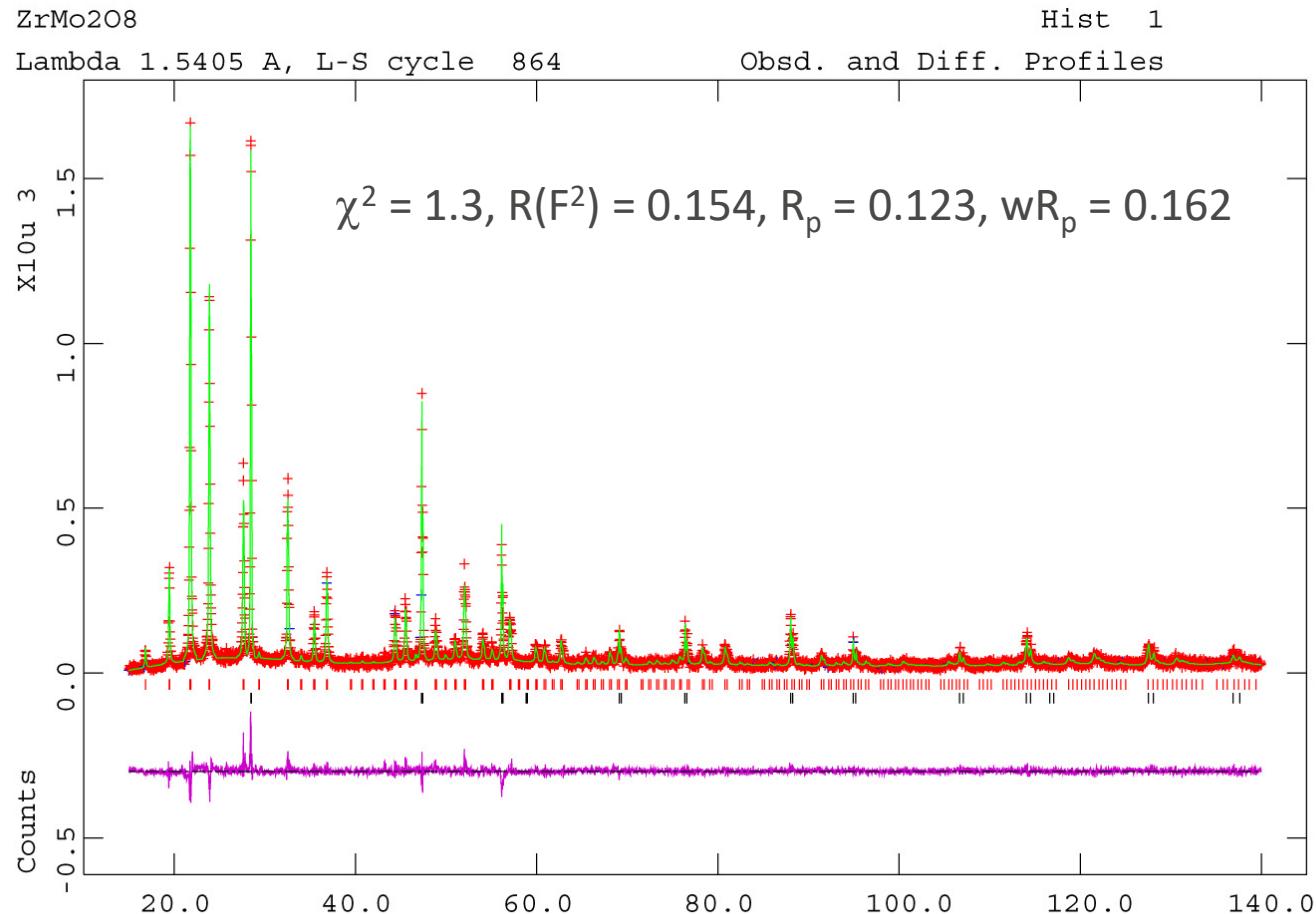
- $R_w = \frac{\sum |y_i(obs) - y_i(calc)|}{\sum y_i(obs)}$

- $R_F = \frac{\sum |(I_K(obs))^{1/2} - (I_K(calc))^{1/2}|}{\sum (I_K(obs))^{1/2}}$

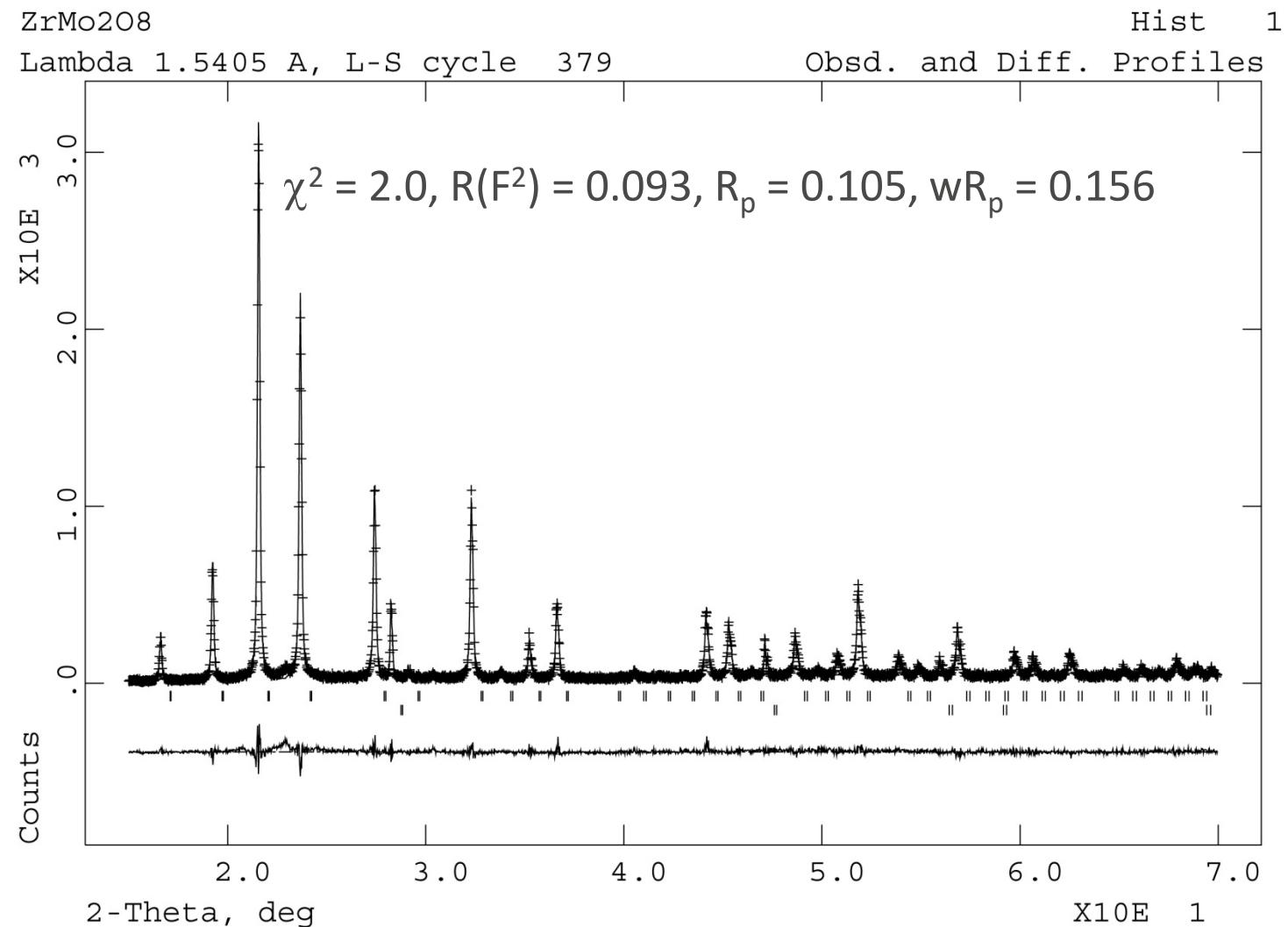
- Goodness of fit Chi<sup>2</sup>

- The easiest way to get low Chi<sup>2</sup> values is to collect noisy data
- $R_F$  is not minimized during the refinement, but gives information about the agreement between the structural model and the pattern
- The most important judgment is in all cases the visual judgment!
  - Low indices mean nothing if the fit does not look convincing

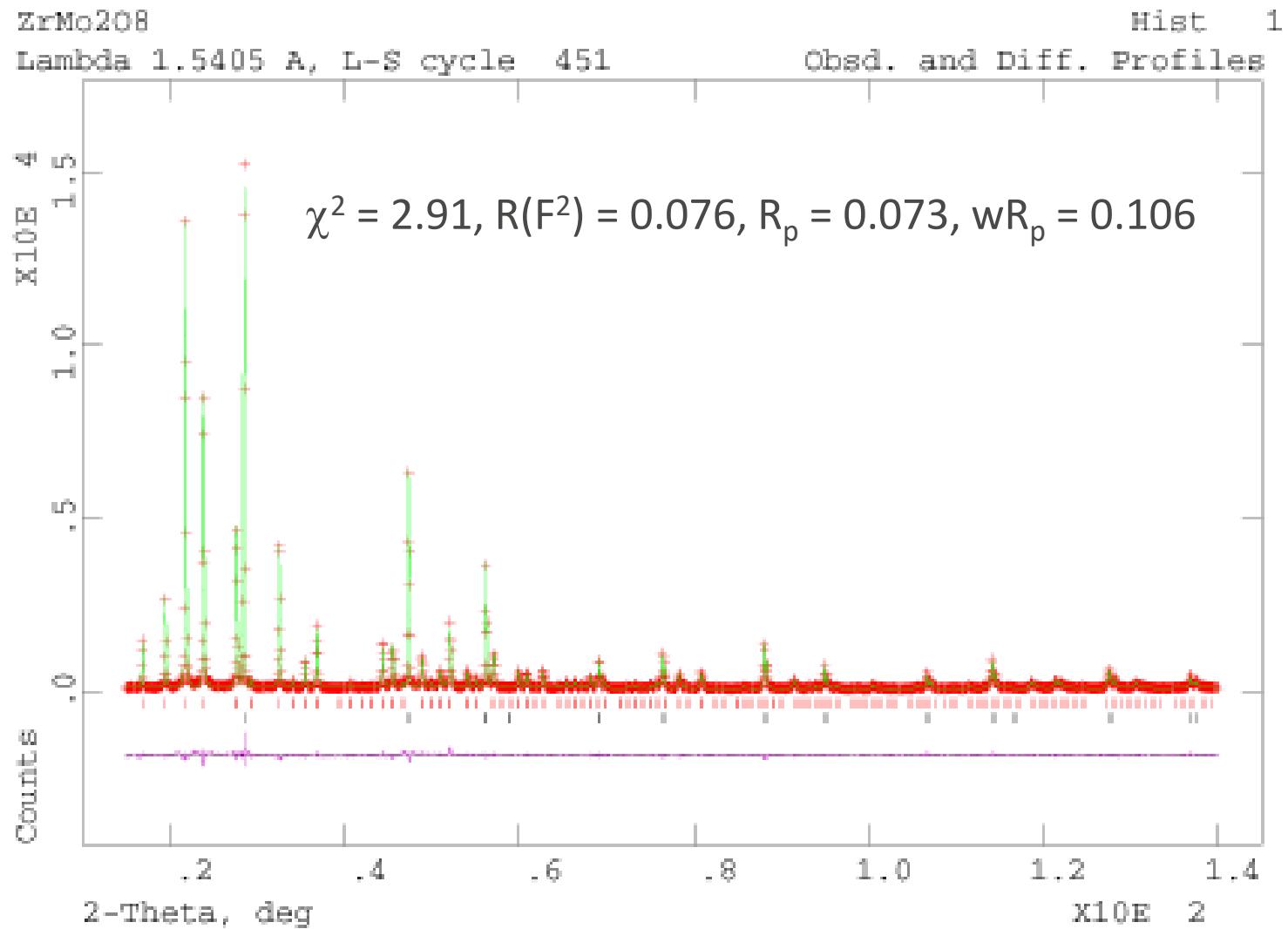
# A good fit



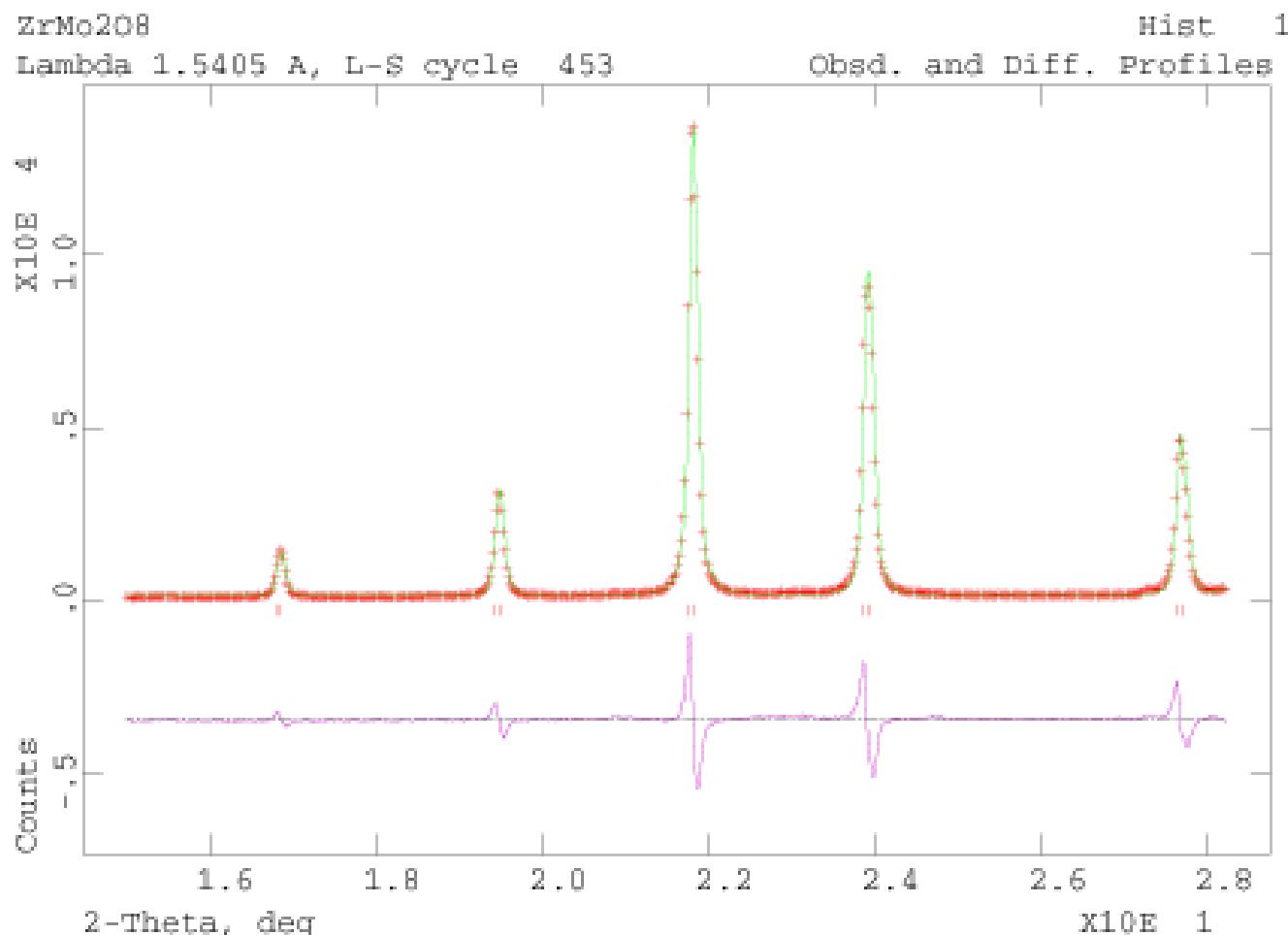
# Also a good fit



# Another good fit



# A fit that is clearly not there yet



$$\chi^2 = 8.54, R(F^2) = 0.115, R_p = 0.139, wR_p = 0.182$$

Peaks are shifted due to incorrect lattice constant

# Modern instrumentation

- **High resolution diffractometers and synchrotron data can provide extra information**
  - Better peak resolution (FWHM can be as low as  $0.005^\circ$  in some cases)
  - High peak to background ratios
- **However, there is a price to pay for this:**
  - Peak shape often determined by the sample
  - TOF diffractometers have asymmetric peaks
- **Modeling of high-resolution data can be very demanding**

# Refining several patterns simultaneously

- **Refinement of several patterns simultaneously can provide additional information**
- **X-ray and neutron data**
  - X-rays usually give better resolution, but the intensity falls off at small d-spacings
  - Good for resolving small lattice distortions
  - Neutron data provide much higher intensities at small d-spacings
  - More reliable atomic positions and thermal parameters, esp. for light atoms
- **X-ray data of several different wavelengths**
  - Can be used to change the scattering contrast between atoms
  - More reliable refinement of partial occupancies

# What to watch out for

- **Some parameters can be highly correlated**
  - Unit cell constants and wavelength
  - Thermal parameters and partial occupancy
  - Sample height and zero point
  - Look at correlation matrix!

- **Check whether your results are physically meaningful!**

- If you know the wavelength, fix it
- Does your occupancy agree with your chemical composition?
- Bond lengths?

- **Use an internal standard if possible**

- Known lattice constant
- Constrain sample height, zero-point, asymmetry etc.  
to be identical for standard and sample



# Final steps

- **Closely inspect your fit/difference curve**
- **Any bad correlations?**
  - Look at correlation matrix!
- **Check whether your results are physically meaningful!**
  - If you know the wavelength, fix it
  - Does your occupancy agree with your chemical composition?
    - Make sure you take into account multiplicity of positions!
  - ADPs from laboratory powder data are often not very meaningful, so don't stress too much if that is the only problem
  - But bond distances and packing must be reasonable!

# Summary: Some basic “rules of thumb” for Rietveld refinement

- Collect good data
- Your eyes are the most important judgement tool: Inspect the difference curve!
- If you refine several phases, constrain changes in parameters like sample height or asymmetry to be identical for all phases before varying them!
- If you are using an internal standard, do NOT vary its lattice constants!
- Suggested order for turning on parameters unless you spot a “biggest culprit”:
  - automatically varied in GSAS: 3 background parameters, histogram scale
  - if necessary, insert more background parameters
  - lattice constants (except for standard; make sure the starting point isn't too far off!)
  - profile parameters LX, LY, SHFT, ASYM; if necessary, also GW, TRNS
  - ADPs (keep isotropic, possibly constrain), atom positions
  - preferred orientation, diffractometer constants, absorption, extinction etc. should only be varied if absolutely necessary!

# Useful resources

- CCP14: Free software including tutorials and examples  
<http://www ccp14.ac.uk/>
  - Unfortunately, no longer maintained due to lack of funding
- Rietveld mailing list  
<http://crystsun1.unige.ch/stxnews/riet/welcome.htm>
- GSAS-II download  
<https://subversion.xray.aps.anl.gov/trac/pyGSAS>
- R. A. Young; “The Rietveld method”
  - Comprehensive text including history, description of several Rietveld programs, as well as details about certain parameters (e.g., background modeling, peak shapes, pattern decomposition...)